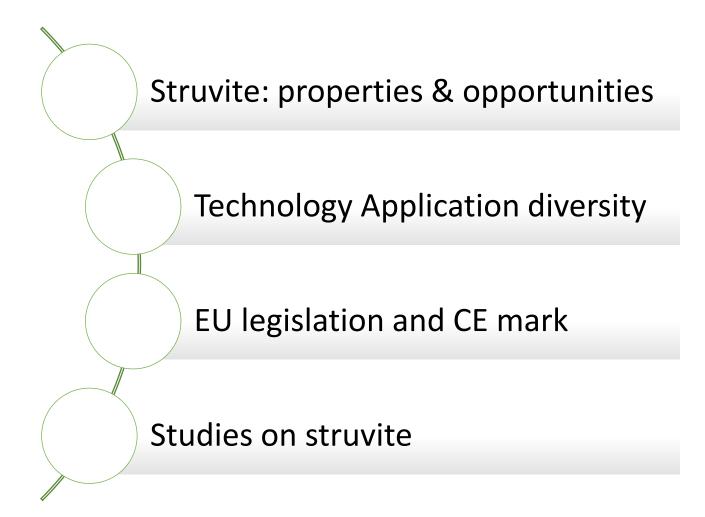


Struvite – a new option for improving phosphorus balances in organic agriculture



Table of contents





Struvite as an Alternative Fertilizer

- Mineral fertilizer = Limited resource & high environmental impact No P mines in EU (Norway ?) Phosphorus listed as critical substance EU
- Quality concerns depending on Mg reagent used (soluble versus low soluble Mg salts)
 Lower quality but consistent product.
- Struvite = excellent fertilizer properties but bad fertilizer LOW SALINITY = Droughts

MgNH₄PO₄.6H₂O 5,7% N 12,6% P 10% Mg 0% K

 No K = but opportunity at hand in producing K-struvite = still bad NPK ratio

MgKPO₄.6H₂O 0% N 11,4% P 8,7% Mg 17,7% K

- Recent research = struvite versus mineral = at least as good or even better
 - = clear advantage (NO) emission = due to slow-release properties



Slow-release fertiliser

Can be used as a precision fertilizer and in organic farming:

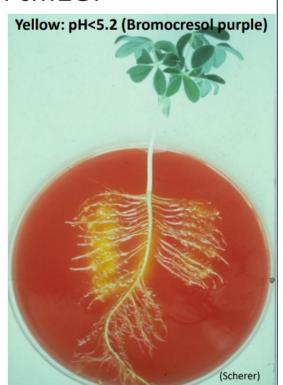


Struvite is a Controlled Release Fertilizer

- Roots "seek" for nutrients in soils (this is true for N, P).
- Roots concentrate around the struvite pellet.
- By acidification, the roots solubilize the struvite and take nutrients up efficiently.

Struvite pellet after 4 months







Struvite as an Alternative Fertilizer

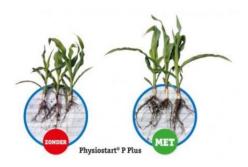
 Root induced organic acid release Nitrification by soil microbiota



https://crystalgreen.com/

• Phyto-start (Timac)





Need for Market-pull product





Struvite as a 'feed material'

Further processing of struvite can add value.



How to turn Struvite into a Fertilizer Product

Pelletizer







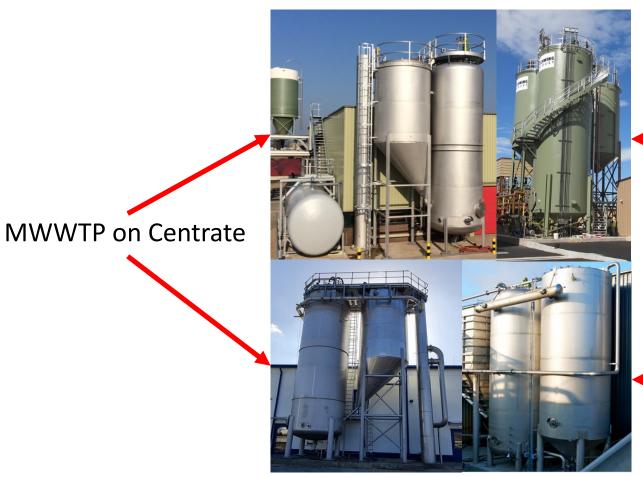
Granulator







Application diversity



MWWTP on Digestate

IWWTP on Centrate

















Technology – General

Motivation P recovery ≠ P recovery as fertilizer

Scaling Prevention & Gritt clogging reactors



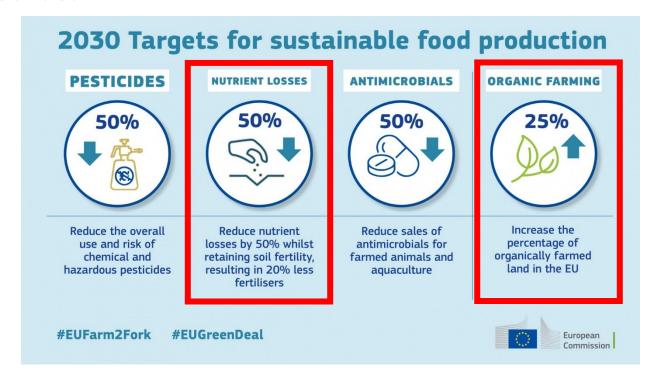
- Reducing back-flow PO4-P (up to 25% of P load)
- Stabilizing Bio-P removal in main WWTP



Depending on the specific situation at your plant and your immediate and/or future expectations, NuReSys combines its different technologies in order to Integrate Phosphate Management solutions.

Benefit resulting from using Struvite

EU Greendeal



GENERAL:

Reduced GHG emmission @ high level NOx Decreased dependency of imported phosphates Transition to circular economy



Regulation 2019/1009 - EU fertilizers

An EU fertiliser product

Struvite

Requirements of product function category (Annex I)

PFC 1(C)(I)(a)(ii): COMPOUND SOLID INORGANIC MACRONUTRIENT FERTILISER

Minimum concentrations*

- 3 % of total nitrogen (N),
- 3 % of total phosphorus pentoxide (P2O5),
- 3 % of total potassium oxide (K2O),
- 1,5% of total magnesium oxide (MgO),
- 1,5 % by mass of total calcium oxide (CaO),
- 1,5 % by mass of total sulphur trioxide (SO₃), or
- 1% of total sodium oxide (Na 2 O).

Total sodium oxide (Na 2 O) < 40 %.

The sum of all declared macronutrient contents shall be at least 18 % by mass.

Requirements of relevant component material category (Annex II)

CMC 12: PRECIPITATED PHOSPHATE SALTS AND DERIVATES

Struvite recovered from wastewater is allowed (both municipal and industrial)

The precipitated phosphate salts shall contain:

- phosphorus pentoxide (P2O5) > 16 % of the dry matter content;
- Organic carbon (Corg)< 3 % of the dry matter content;
- <3 g/kg dry matter of macroscopic impurities above
 2 mm (organic matter, glass, stones, metal and plastics;
- <5 g/kg dry matter of the sum of the macroscopic impurities referred.

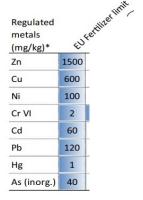


EU Regulation - CMC12 conti.

- The precipitation process shall take place under controlled conditions in a reactor.
- Pathogens:

Minn annuisment he total	Sa	ampling pla	ns	Limit					
Micro-organisms to be tested	n	С	m	М					
Salmonella spp.	5	0	0	Absence in 25 g or 25 ml					
Escherichia coli or Enterococcaceae	5	5	0	1 000 in 1 g or 1 ml					
Clostridium perfringens	5	5	0	100 CFU in 1 g or 1 ml					
Ascaris sp. viable eggs	5	0	0	Absence in 25 g or 25 ml					

- The dry matter of precipitated phosphate salts and derivates shall be measured using vacuum drying at 40 °C until constant weight to avoid the loss of crystal-bound water.
- Heavy metal contamination:





EU FPR-CE Mark

- The regulatory framework is in place EU 2019/1009 and EU 2019/1009
- Notified body can examine each production site to see if struvite meets the laid-out criteria
- EU Fertiliser product regulation CE mark can be obtained
- ✓ Allows use in organic farming



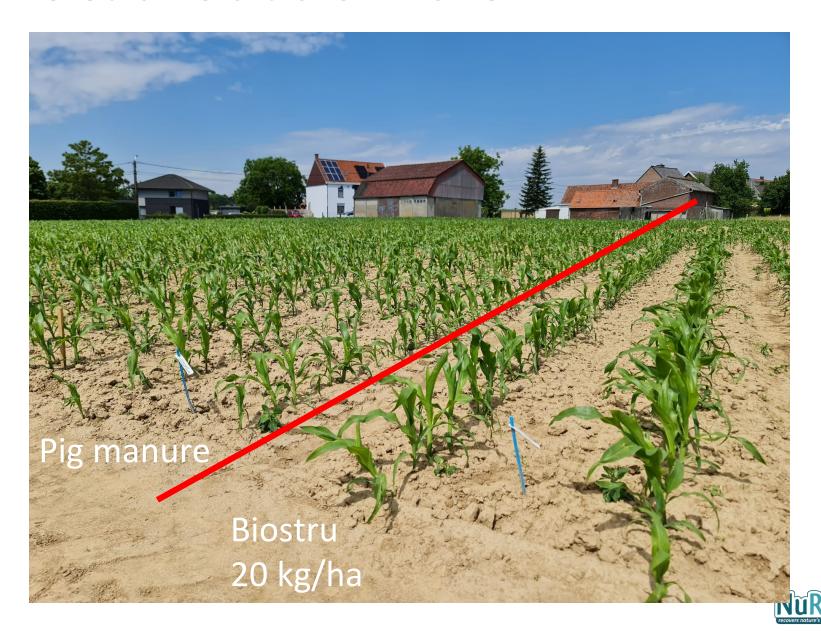
Application of struvite as a fertilizer

- Used as a booster in the sowing stage
- Used in organic farming
- Mixed with conventional fertilisers to optimise NPK
- Mixed with organic/recovered fertilisers (FERPLAY)



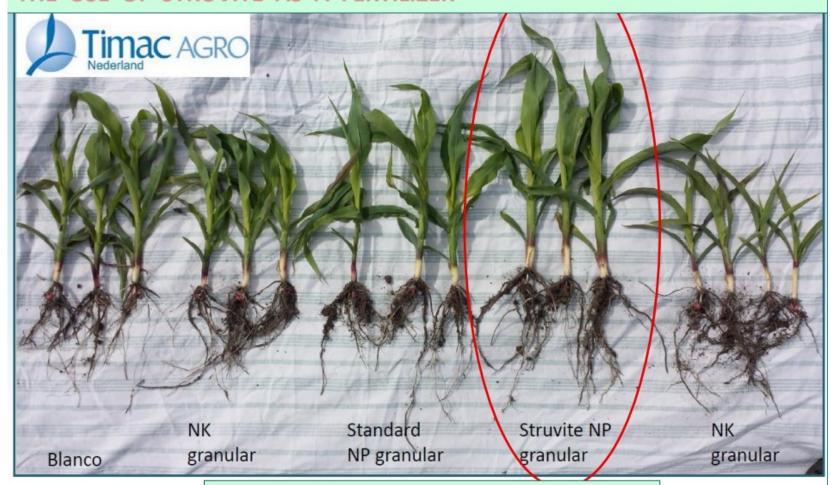


Bio-Stru field trails - Maize



Bio-Stru field trails - Maize

THE USE OF STRUVITE AS A FERTILIZER



TIMAC AGRO has turned the "slow fertilizer" STRUVITE into a STARTER FERTILIZER for maize.



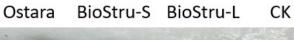
Struvite field trails – far Asia

Ostara BioStru-S BioStru-L CK



- Ostara was purchase from the dealer shop in Taiwan
- Use same gram of Ostara, BioStru-S and –L (2 different particle size) and directly mix with pak choy roots.
- CK, water + foliar fertilizer 14-15-10 1 time on 10 Mar.
- We test 2 weeks, there is no harmful to roots.







A systematic comparison of commercially produced struvite: Quantities, qualities and soil-maize phosphorus availability

Table 3Heavy metal content of the analyzed struvite samples (Empty cells: measured content < limit of detection (LOD). N.A.: Not Applicable). Color fill of the cells indicates to which extend the regulatory limit is approached or if no regulation applies.

		Type I: digestate									Type II: dewatering liquor of digestate											Type III: UASB effluent						
Regulated metals (mg/kg)*	ξÚ	fertilize	P limit	rpet 2	irpret?	ipret3	TOPE A	in prets	HAY 2	JRESYS VE	cht or	105greet	10 Steel	2 105greet	JRESYS T	JRESYS?	nohos 1	iphos 2	earl's	earl 2	2313 4	aske N	iresys A	iresys 5	10503Q	iphos 3	iphos a Anphos	
Zn	1500	1	50	24	40	26	33	2	13	33		4.2	1.6	1.8	55		5.6	1.4			13	8.1	14	548	241	501	4.8	
Cu	600	1	28	17	16	7.4	7.5	2	3.3	20					64		1.5				6.3	1.2	1.4	70	73	92		
Ni	100	1	2.3	3.4	1.5	1.4	1.1			1.3					4						12			5.5	6	5.2		
Cr VI	2	1.5																										
Cd	60	1																							1.9	2		
Pb	120	2	19	2.4	3.4	3.4				3.5					9.9									3.7	3.1	2.2		
Hg	1	0			0		0.1			0					0						0							
As (inorg.)	40	0.5	0.6												0.6						5.7			1.3	1	1.2		
Metals not	regula	ited f	or in t	he fe	rtilize	r regu	lation	n (mg/	/kg)																			
Mn	N.A.	1	179	129				30	193		43	64	94	28	441	28	688	211				121		201			154	
Cr	N.A.	1	2.9	1.6	3.3	4.8	2	6.2	7	2.5				6.8	13						2.7	7.5		9	5.4	5.7		



A systematic comparison of commercially produced struvite: Quantities, qualities and soil-maize phosphorus availability

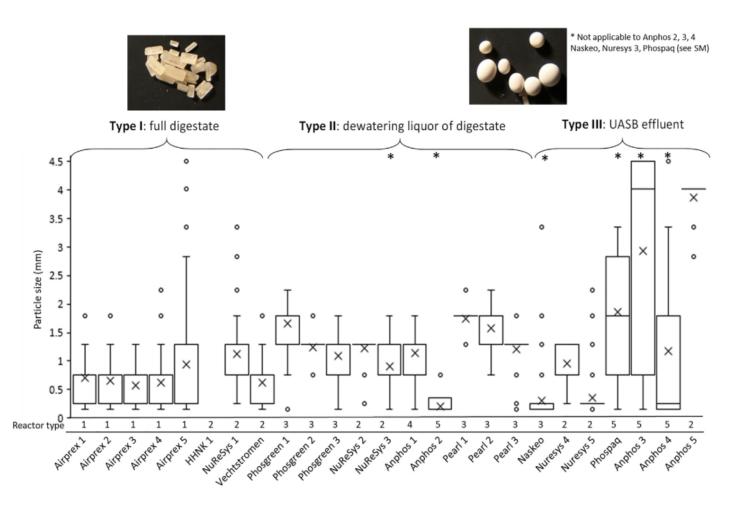


Fig. 5. Mass fraction distribution per particle size for all analyzed struvite samples. Crosses indicate the mean particle size (mm). Numbers indicate the reactor type as follows: 1 Airlift reactor, 2 Continues stirred tank reactor, 3 Fluidized bed reactor, 4 Tank aerated, 5 Tank mixed. Histograms, descriptive statistics and pictures per sample are presented in the SM Section 6, raw data shown in SM Table 4.



A systematic comparison of commercially produced struvite: Quantities, qualities and soil-maize phosphorus availability

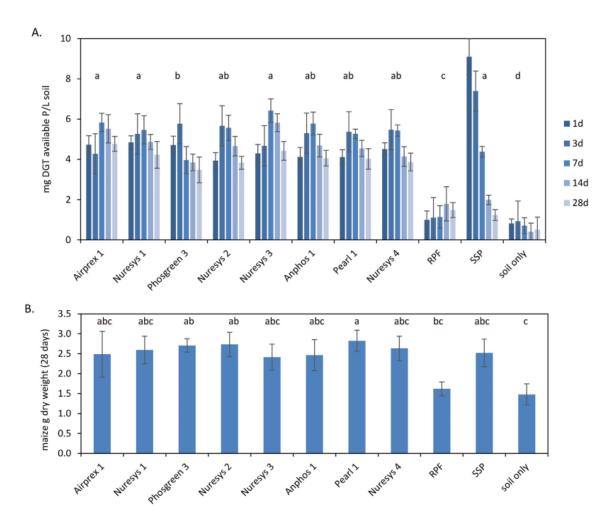


Fig. 6. A) Available P in days after germination in maize seedlings pots fertilized with struvite, rock phosphate (RP), single super phosphate (SSP) or without additional P fertilization (soil only). P collected in 24 h hours with DGT after the indicated time points are shown. P application was normalized to the P content of each sample so that in each pot the same amount of P was applied. B) Maize seedling dry weight after 28 days. Letter codes indicate significant differences between overall release patterns (i.e. different letters represent significant differences p < 0.01), for example letter combinations containing the latter 'a' are not significantly different from each other, while a combination not containing the letter 'a' is significantly different.

Publications for net Struvite Nox emissions

- Yang, Z., Ferron, L. M., Koopmans, G. F., Sievernich, A., & van Groenigen, J. W. (2023). Nitrous oxide emissions after struvite application in relation to soil P status. *Plant and Soil*, 1-15.
- Wang, L., Ye, C., Gao, B., Wang, X., Li, Y., Ding, K., Li, H., Ren, K., Chen, S., Wang, W. and Ye, X., 2023. Applying struvite as a N-fertilizer to mitigate N2O emissions in agriculture: Feasibility and mechanism. Journal of Environmental Management, 330, p.117143.
- Fukumoto, Y., Suzuki, K., Kuroda, K., Waki, M. and Yasuda, T., 2011. Effects of struvite formation and nitratation promotion on nitrogenous emissions such as NH3, N2O and NO during swine manure composting. Bioresource technology, 102(2), pp.1468-1474.
- Britton, A.T., Sacluti, F., Oldham, W.K., Mohammed, A., Mavinic, D.S. and Koch, F.A., 2007, June. Value from waste—struvite recovery at the city of Edmonton's gold bar WWTP. In Proc. IWA Specialist Conference: Moving Forward—Wastewater biosolids sustainability.



